## COMPONENTS:

- (1) 2,4-Dichlorophenol; C<sub>6</sub>H<sub>4</sub>Cl<sub>2</sub>O; [120-83-2]
- (2) Water; H<sub>2</sub>O; [7732-18-5]

### **EVALUATOR:**

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November 1979.

## CRITICAL EVALUATION:

The solubility of 2,4-dichlorophenol in water was determined many decades ago by Mosso (1). While the measurements may be subject to question because of their age and obscurity, the reported melting point strongly indicates that the "m-dichlorophenol" does refer to 1,2-dichlorophenol. Undoubtedly, the experimental procedure was based upon a classical gravimetric analysis. Because of the lack of other data, the value recommended here must be classified as doubtful.

The following solubility value for 2,4-dichlorophenol in water is based solely upon the work reported by Mosso.

T/K	$10^2$ mol(1)/dm $^3$	g(1)/kg	$10^4 x(1)$
293.15	2.7	4.5	4.9

#### REFERENCES

1. Mosso, U. Jahresber. Fortschr. Chem. 1887, 1800.

# COMPONENTS: ORIGINAL MEASUREMENTS: (1) 2,4-Dichloropheno1; C<sub>6</sub>H<sub>4</sub>Cl<sub>2</sub>O; [120-83-2] Mosso, U. Jahresber. Fortschr. Chem. 1887, 1800. (2) Water; H<sub>2</sub>O; [7732-18-5] VARIABLES: PREPARED BY: A. Vesala One temperature EXPERIMENTAL VALUES: t/°C g(1)/kg(2) a $10^2$ mol(1)/kg b $10^4$ x(1) b 20 4.47-4.66 2.730-2.845 4.938-5.148 a. Reported.b. Calculated by F. W. Getzen. AUXILIARY INFORMATION SOURCE AND PURITY OF MATERIALS: METHOD/APPARATUS/PROCEDURE: $C_6H_4C1_2O$ : Synthesized from dry chlorine Experimental methods were not described. and phenol, distilled and recrystallized from benzene, reported melting point 43 - 44°C. н,о: Source and purity not specified. ESTIMATED ERROR: REFERENCES: